A Facile Route to 4,5,6,7-Tetrahydrothiophenes and Benzothiophenes Takehiko Nishio*, Norikazu Okuda and Choji Kashima

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2-Substituted tetrahydro- and cyclopentathiophenes 2 were produced in high yields by treatment of the semi-cyclic trisubstituted 1,4-diketones 1 such as 2-phenacylcyclohexan-1-one, 2-acetonylcyclohexan-1-one, and 2-phenacylcyclopentan-1-one, with Lawesson's reagent.

J. Heterocyclic Chem., 25, 1437 (1988).

2-Bis(p-methoxyphenyl)-1,3-dithiaphosphophetane-2,4-disulfide (Lawesson's reagent; LR) is known to be a most effective thionation reagent for a wide variety of carbonyl compounds [1]. It is reported that treatment of bifunctional substrates with LR undergoes ring closure to form heterocycles [2]. Recently, Shridhar et al. [3] reported that the interaction of tri- and tetra-substituted butane-1,4-diones with LR resulted in the formation of a mixture of the thiophenes and furans in varying proportion, while that of di-substituted butane-1,4-diones with LR produced the thiophenes exclusively. We now report that semi-cyclic 1,4-diketones 1, which are regarded as tri-substituted butane-1,4-diones, reacted with LR to give the thiophene derivatives 2 exclusively in high yields.

A mixture of semi-cyclic 1,4-diketones la-e,g and equimolar of Lawesson's reagent (LR) in a mixed benzene-dimethoxyethane solution was heated under reflux to give 2-substituted 4,5,6,7-tetrahydrothiophenes 2a-e and cyclopentathiophene (2g) in 58-99% yields. The structure of the products 2 was elucidated on the basis of spectral data and elemental analyses. The yield of 2-phenyl-4,5,6,7-tetrahydrothiophene (2a) decreased to about ½ by use of 0.5 mole of LR per mole of 2-phenacyl-cyclohexan-1-one

(1a). Consequently, the reaction presumably involves the formation of 1,4-dithioxo intermediate which undergoes cyclization and elimination of hydrogen sulfide to give 2. 2-Phenyl- (2a) and 2-methyl-4,5,6,7-tetrahydrothiophene (2e) thus obtained were easily oxidized with chloranil or 2,3-dichloro-5,6-dicyano-p-benzoquinone (DDQ) to give the corresponding 2-substituted benzothiophenes 3a,e. On the other hand, treatment of 4-oxo ester 1f with LR gave an intractable mixture.

In conclusion, the reaction described here would be a facile method for the synthesis of 2-substituted 4,5,6,7-tetrahydrothiophenes 2 and benzothiophenes 3.

EXPERIMENTAL

The Reaction of 1,4-Diketones la-g with Lawesson's Reagent.

A solution of the 1,4-diketone (1, 1 mmoles) and Lawesson's reagent (LR, 1.1 mmoles) in benzene-dimethoxyethane (20-10 ml) was heated under reflux for 1 hour. After removal of the solvent in vacuo, the residual oil was chromatographed on a silica gel column (Wakogel C-300, flash chromatography) with benzene-hexane (1:4) as eluant to yield the corresponding tetrahydro-, 2a-e, and cyclopentathiophenes 2g.

aD.5 eq. molar of LR was used. bDDQ was used as an oxidizing agent

CIntractable mixture.

Table 1

The Yield and NMR Spectral Data of Tetrahydro- and Cyclopentathiophenes 2 and Benzothiophenes 3

| | Yield (%) | 'H-NMR (Deuteriochloroform) (δ) | ¹³ C-NMR (Deuteriochloroform) (δ) |
|------------|-------------|--------------------------------------------------------------------------------------------------------------------------|------------------------------------------------------------------------------------------------------------------------------------------|
| 2a | 91 (47) [a] | 1.60 (4H, m), 2.40-2.90 (4H, m), 6.95 (1H, s), 7.05-7.66 (5H, m) | 22.9 (t), 23.6 (t), 25.1 (t), 25.6 (t), 123.7 (d), 125.4 (d), 126.9 (d), 128.7 (d), 134.8 (s), 135.5 (s), 136.3 (s), 140.3 (s) |
| 2b | 99 | 1.62-1.94 (4H, m), 2.32 (3H, s), 2.44-2.82 (4H, m), 6.90 (1H, s), 7.11 (2H, d, $J = 7.8$ Hz), 7.42 (2H, d, $J = 7.8$ Hz) | 21.1 (q), 22.9 (t), 23.6 (t), 25.0 (t), 25.6 (t), 123.2 (d), 125.4 (d), 129.3 (d), 132.1 (s), 134.8 (s), 136.1 (s), 136.6 (s), 140.5 (s) |
| 2c | 88 | 1.65-1.95 (4H, m), 2.50-2.82 (4H, m), 3.78 (3H, s), 6.87 (1H, s), 6.85 (2H, d, J = 8.8 Hz), 7.45 (2H, d, J = 8.8 Hz) | 22.9 (t), 23.7 (t), 25.0 (t), 25.6 (t), 55.3 (q), 114.1 (d), 122.7 (d), 126.7 (d), 127.8 (s), 134.4 (s), 136.2 (s), 140.2 (s), 158.8 (s) |
| 2d | 78 | 1.60-1.98 (4H, m), 2.38-2.84 (4H, m), 6.91 (1H, s), 7.21-7.58 (4H, m) | 22.9 (t), 23.6 (t), 25.1 (t), 25.6 (t), 124.1 (d), 126.5 (d), 128.8 (d), 132.5 (s), 133.4 (s), 136.0 (s), 136.5 (s), 138.9 (s) |
| 2e | 67 | 1.61-1.95 (4H, m), 2.24-2.75 (4H, m), 2.40 (3H, s), 6.39 (1H, s) | 15.5 (q), 23.1 (t), 23.8 (t), 24.9 (t), 25.6 (t), 125.8 (d), 133.2 (s), 135.0 (s), 135.8 (s) |
| 2g | 58 | 2.25-2.97 (6H, m), 7.02 (1H, s), 7.00-7.95 (5H, m) | 28.4 (t), 29.0 (t), 29.3 (t), 118.4 (d), 125.3 (d), 126.9 (d), 128.7 (d), 135.5 (s), 141.8 (s), 147.2 (s) |
| 3a | 91 (79) [b] | 7.16-7.92 (10H, m) | 119.4 (d), 122.2 (d), 123.5 (d), 124.3 (d), 124.4 (d), 126.4 (d), 128.2 (d), 128.9 (d), 134.3 (s), 139.1 (s), 140.6 (s), 144.2 (s) |
| 3 e | 41 | 2.53 (3H, d, $J = 1.5$ Hz), 6.93 (1H, br d, $J = 1.5$ Hz), 7.09-7.39 (2H, m), 7.53-7.78 (2H, m) | 16.1 (q), 121.6 (d), 122.0 (d), 122.5 (d), 123.3 (d), 124.0 (d), 139.7 (s), 140.5 (s), 140.8 (s) |

[a] 0.5 equimolar of LR was used. [b] DDQ was used as oxidizing agent.

2-Phenyl-4,5,6,7-tetrahydrothiophene (2a).

This compound had mp 79-80° (lit [4] 82.5-83.5°); ir (potassium bromide): 1590, 1500, 1440, 835, 760, 700 cm⁻¹.

2-p-Methylphenyl-4,5,6,7-tetrahydrothiophene (2b).

This compound had mp 63-64°; ir (potassium bromide): 1505, 1440, 1105, 815, 805 cm⁻¹.

Anal. Calcd. for C₁₅H₁₆S: C, 78.89; H, 7.06. Found: C, 79.15; H, 7.13.

2-p-Methoxyphenyl-4,5,6,7-tetrahydrothiophene (2c).

This compound had mp 117-118°; ir (potassium bromide): 1605, 1515, 1440, 1245, 1035, 825 cm⁻¹.

Anal. Calcd. for C₁₅H₁₆OS: C, 73.73; H, 6.60. Found: C, 74.06; H, 6.65.

2-p-Chlorophenyl-4,5,6,7-tetrahydrothiophene (2d).

This compound had mp 122-123°; ir (potassium bromide): 1495, 1435, 1085, 1000, 820, 810 cm⁻¹.

Anal. Calcd. for C₁₄H₁₃ClS: C, 67.59; H, 5.26. Found: C, 67.71; H, 5.29. 2-Methyl-4,5,6,7-tetrahydrothiophene (2e).

This compound had bp 85°/2 mm Hg (lit [5] 110-111°/19 mm Hg); ir (film): 1440, 1200, 1120, 830, 810 cm⁻¹.

2-Phenylcyclopentathiophene (2g).

This compound had mp 65-65.5°; ir (potassium bromide); 1590, 1490, 835, 750, 690 $\rm cm^{-1}$.

Anal. Calcd. for C₁₈H₁₂S: C, 77.95; H, 6.03. Found: C, 78.01; H, 6.09.

Oxidation of 2-Substituted 4,5,6,7-Tetrahydrothiophenes 2a,e.

A mixture of the 4,5,6,7-tetrahydrothiophene 2 (200 mg) and Chloranil

(or DDQ) (2.5 equimolar) in toluene (30 ml) was refluxed under argon for 20 hours. An usual work-up as described above gave 2-substituted benzothiophene 3.

2-Phenylbenzothiophene (3a).

This compound had mp 170-171° (lit [6] 170-172°): ir (potassium bromide) 1480, 1440, 820, 750, 720, 680 $\rm cm^{-1}$.

2-Methylbenzothiophene (3e).

This compound had mp 50° (lit [7] 51-52°): ir (potassium bromide) 1455, 1430, 1200, 830, 745, 725 cm $^{-1}$.

REFERENCES AND NOTES

- [1] M. P. Cava and M. I. Levinson, Tetrahedron, 41, 5061 (1985).
- B. S. Pederson and O. S. Lawesson, *ibid.*, 35, 2433 (1979); A. A. El-Barbary, S. Scheibye, O. S. Lawesson and H. Fritz, *Acta. Chem. Scand.*, B34, 597 (1980); A. A. El-Barbary, K. Clausen and S. O. Lawesson, *Tetrahedron*, 36, 3309 (1980).
- [3] D. R. Shridhar, M. Jogibhukta, P. S. Rao and V. K. Handa, Synthesis, 1061 (1982); idem., Indian J. Chem., 1187 (1983).
 - [4] A. W. Horton, J. Org. Chem., 14, 761 (1949).
- [5] S. H. Groen, R. M. Kellogg, J. Buter and H. Wynberg, J. Org. Chem., 33, 2218 (1968).
- [6] K. E. Schulte, J. Reisch and D. Bergenthal, Chem. Ber., 101, 1540 (1968).
 - [7] R. T. Dickinson and B. Iddon, J. Chem. Soc. (C), 2733 (1968).